Cellular to Multicellular 6th Grade Learning Targets

Content Statement:

Cells are the fundamental unit of life.

I can...

- 1. Describe the basic modern cell theory.
- 2. Identify and explain that all living things have **six** basic characteristics: 1 Made of cells, 2 respond to stimuli, 3 grow and develop, 4 reproduce (asexual or sexual), 5 adapt, and 6 obtain and use energy.
- 3. Identify and explain the parts and functions of a microscope.
- 4. Identify the basic organelles of a plant cell: cell wall, cell membrane, cytoplasm, vacuole, chloroplasts, mitochondria, ribosome, nucleus, chromosomes.
- 5. Identify the basic organelles of an animal cell.

Vocabulary:

Content Statement:

All cells come from pre-existing cells.

I can...

- 1. Identify and organize the phases of mitosis.
- 2. Understand how cells transmit genetic information from parent cell to daughter cell.
- 3. Order the phases of mitosis for a plant.
- 4. Order the phases of mitosis for an animal.
- 5. Compare and contrast the differences between plant and animals cell undergoing mitosis.
- 6. Describe how mitosis disproved the misconception of spontaneous generation.
- 7. Model the movement of chromosomes in mitosis.

Vocabulary:

Mitosis Parent cell Daughter cell Chromosome	Spontaneous generation	
CHIOMOSOME		

Content Statement:

Cells carry on specific functions that sustain life.

I can...

- 1. Explain the function of each of the organelles in a plant and animal cell.
- 2. Explain how all of the parts work together to keep the cell alive.
- 3. Explain the role of cells, tissues, organs, and organ systems in an organism
- 4. Explain that cells form tissues.
- 5. Explain that tissues form organs.
- 6. Explain that organs form organ systems.
- 7. Explain that specialized cells have different functions (such as, homeostasis, gas exchange, energy transfers, transportation of molecules, disposal of wastes, and synthesis of new molecules).
- 8. Explain the relationship between cells, tissues, organs, and organ systems.
- 9. Illustrate how specialized cells form tissues, organs, and organ systems.
 - **Note, photosynthesis is covered in 5th grade science.

Vocabulary:

Cells	Homeostasis	Osmosis
Tissues	Gas exchange	Diffusion
Organs Organ systems	Energy transfer	Permeable Semi-permeable Impermeable

Content Statement:

Living systems at all levels of organization demonstrate the complementary nature of structure and function.

I can...

- 1. Define multicellular.
- 2. Identify a variety of specialized cells, tissues, organs, and systems that perform specialized functions in multicellular organisms.
- 3. Identify organisms that have various forms of symmetry (i.e., radial symmetry and bilateral symmetry).
- 4. Recognize the difference between vertebrate and invertebrate organisms.
- 5. Identify the different systems in multicellular organisms (respiratory, circulatory, digestive, skeletal, muscular, etc.)
- 6. Describe the function of the different systems in multicellular organisms (respiratory, circulatory, digestive, skeletal, muscular etc.).
- 7. Describe the purpose of the different systems in multicellular organisms (respiratory, circulatory, digestive, skeletal, muscular, etc.)
- 8. Identify and label the parts of the body systems by dissecting multicellular organism(s).
- 9. Compare and contrast the functions of the body systems of vertebrates and invertebrates.
- 10. Define the three types of symmetry (radial, bilateral, and asymmetry).
- 11. Compare and contrast the complexity of body systems in the different symmetries.
- 12. Compare the four major types of tissues (epithelial, connective, nerve and muscles tissue).

Vocabulary:

Multicellular	Tissue	
Symmetry	Epithelial tissue	
Radial symmetry	Connective tissue	
Bilateral symmetry	Nerve tissue	
Asymmetry	Muscle tissue	
Organ system	0.5 2.5 2.5 0.5 0.5 0.5 0.5 0.5 0.5 0.5 0.5 0.5 0	

Matter and Motion 6th Grade Learning Targets

Content Statement:

All matter is made up of small particles called atoms

I can...

- 1. Explain that all matter is made of atoms (atomic theory)
- 2. Define an element; know that all substances are made of different elements
- 3. Molecules are the combination of 2 or more atoms or compounds joined together chemically
- 4. Compounds are composed of 2 or more different elements.
- 5. There are ~90 naturally occurring elements
- 6. Each compound has its own unique, unchanging composition of type and number of elements and atoms.
- 7. Differentiate among atoms, elements, molecules and compounds
- 8. Classify examples of atoms, elements, molecules, and compounds
- 9. All particles (atoms and/or molecules) of a pure substance have nearly identical mass.
- 10. Particles of different substances usually have different masses, depending on their atomic composition.
- 11. Density is the ratio of mass to volume.
- 12. Know that while mass and volume of a material can change but the density generally remains constant.
- 13. Use the mass and volume of different substances to compare their relative densities.
- 14. Compare equal volumes of different substances and know the one that weighs more is denser.
- 15. Use density to identify a material.
- 16. Construct and mass vs. volume graphs
- 17. Interpret mass vs. volume graphs

Vocabulary:

Matter	Density	
Atoms	Mass	
Element	Volume	
Molecules		
Compounds		

Content Statement:

Changes of state are explained by a model of matter composed of atoms and/or molecules that are in motion

I can...

- 1. Explain when substances undergo changes of state, neither atoms nor molecules themselves are changed in structure.
- 2. Define and give examples of solids, liquids, and gasses.
- 3. Describe the motion and spacing of particles in solids, liquids, gasses.
- 4. Differentiate between temperature and thermal energy
- 5. Describe how an object has kinetic energy due to its motion.

- 6. Thermal energy depends on the amount/mass of the substance whereas temperature does not depend on the amount of the substance
- 7. Discuss the relationship among kinetic energy, thermal energy, and temperature
- 8. During phase changes, matter conserved.
- 9. Explain the motion and energy of particles during phases change.
- 10. Explain the Law of Conservation of Matter.
- 11. Describe phase changes. (melting, evaporating, condensation, freezing)??

Vocabulary:

Solid	Thermal energy	Law of Conservation of Matter
Liquid	Temperature	Melting
Gas		Evaporating
		Condensation
		Freezing

Content Statement:

There are two categories of energy: kinetic and potential.

I can...

- 1. Define potential energy.
- 2. Define kinetic energy.
- 3. Classify the different forms energy, (electrical, thermal, sound, gravitational potential) as potential, kinetic, or a combination of the two.
- 4. Explain the relationship between height and gravitational potential energy.
- 5. Explain the relationship between speed and kinetic energy.
- 1. .

Vocabulary:

Kinetic energy Potential energy	Electrical energy Thermal energy Sound energy Gravitational potential energy	Speed
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Content Statement:

An object's motion can be described by its speed and the direction in which it is moving.

I can...

- 2. Define motion.
- 3. Define force.
- 4. Explain a reference point.
- 5. Explain how motion is detected using reference points.
- 6. Calculate the speed of an object given distance and time.

- 7. Describe average speed.8. Read and interpret graphs of average speed.
- 9. Measure and graph position as a function of time.
- 10. Measure and graph speed as a function of time.11. Analyze graphs to determine position and speed of object.

Vocabulary:

Motion	Average speed	
Force		
Reference Point		

Rocks, Minerals, and Soil 6th Grade Learning Targets

Content Statement:

Minerals have specific, quantifiable properties.

I can...

- 1. Define mineral
- 2. Identify and classify mineral examples using properties such as color, luster, Moh's hardness scale, streak, cleavage, magnetism, flouresence and crystal shape.
- 3. Describe the environments in which minerals are formed and how they are formed in those environments, including:
 - Evaporation (e.g. halite, gypsum, chalk)
 - · Chemical (e.g. calicite, dolomite, kaolinite, corundum)
 - · Igneous (e.g. feldspar varieties, magnetite, varieties of quartz, topaz)
 - Metamorphic (e.g. epidote, talc)

Vocabulary:

Mineral	Luster	Evaporation	
Physical properties	Hardness	Evaporation	
Mohs' Hardness Scale	Cleavage		
	Streak		
	Magentism		
	Flourescence	1	
	Crystal Shape		

Content Statement:

Igneous, metamorphic and sedimentary rocks have unique characteristics that can be used for identification and/or classification.

I can...

- 1. Explain that most rocks are made up of one or more minerals.
- 2. Identify and classify rocks based on the basic characteristics and the distinct environments of the three main types of rocks (igneous, metamorphic, and sedimentary)
 - a. Igneous samples must include varieties of granite, rhyolite, basalt, obsidian, pumice and andesite
 - b. Metamorphic samples must include varieties of schist, gneiss, slate, marble, anthracite and phyllite
 - c. Sedimentary samples must include organic forms, such as coal, coquina, limestone, and chert, and inorganic forms, such as sandstone, shale, conglomerate and breccia
 - d. Some rock samples, such as bituminous coal, coquina and chert may not always fall neatly into one specific rock category

Vocabulary:

Content Statement:

Igneous, metamorphic and sedimentary rocks form in different ways.

I can...

- 1. Explain that some sedimentary rocks are made of organic material (e.g. coal)
- 2. Explain that igneous rocks are formed from cooled magma below ground (intrusive) and lava above ground (extrusive.)
- 3. Explain that metamorphic rocks are formed from other rocks that have been changed by heat and pressure
- 4. Explain that sedimentary rocks are made of particles (sediments) that have been eroded, deposited, compacted, and cemented together
- 5. Apply understanding of the three types of rock to analyze the processes of the rock cycle and how it connects to Ohio
 - a. The typical pattern of coal formation is an important connection to energy in Ohio
 - b. The formation of Ohio sandstone and limestone indicating that a shallow sea once covered Ohio
 - c. Ohio's geologic history and past environmental conditions play an important role in understanding the existing bedrock in Ohio (show maps to assist with this)
- 6. Explain that all rocks can be broken down through the destructive processes of weathering and erosion

Vocabulary:

Intrusive	Weathering	
Extrusive	Erosion	

Content Statement:

Soil is unconsolidated material that contains nutrient matter and weathered rock.

I can...

- 1. Define soil (use the term soil, rather than dirt, they are different)
- 2. Analyze types of soil and how they are formed, based on their properties (texture, color, composition, permeability, porosity, and rate of formation) and their environmental conditions
- 3. Explain that soil forms in horizontal layers, known as horizons
- 4. Use soil sampling equipment to observe and investigate different soil horizons, their properties, and how those properties change
 - a. A connection must be made to environmental conditions, types of bedrock and soil properties.

Vocabulary:

Horizons

Content Statement:

Rocks, minerals and soils have common and practical uses. (The following learning targets should be included throughout the Minerals, Rocks, and Soil Unit or the Project-Based Learning Investigation.)

I can...

- 1. Determine the uses of various minerals, rocks, and soil. Uses of the resources should include:
 - a. construction (e.g., gypsum, metals, gravel, sand, lime, clay), energy (e.g., fossil fuels, radioactive materials)
 - b. transportation (e.g., road salt, asphalt)
 - c. agriculture (e.g., lime, peat, minerals for fertilizers, pesticides),
 - d. domestic use (e.g., metals and gems for jewelry, clay for pottery or sculpting, natural dyes for clothing or paint)
 - e. technology (e.g., lithium, silica)
- 2. Analyze the extraction, use, storage, and disposal of minerals, rocks, and soil
- 3. Describe the conservation of mineral, rock, and soil, which are considered non-renewable resources

Vocabulary:

Extraction	
Conservation	